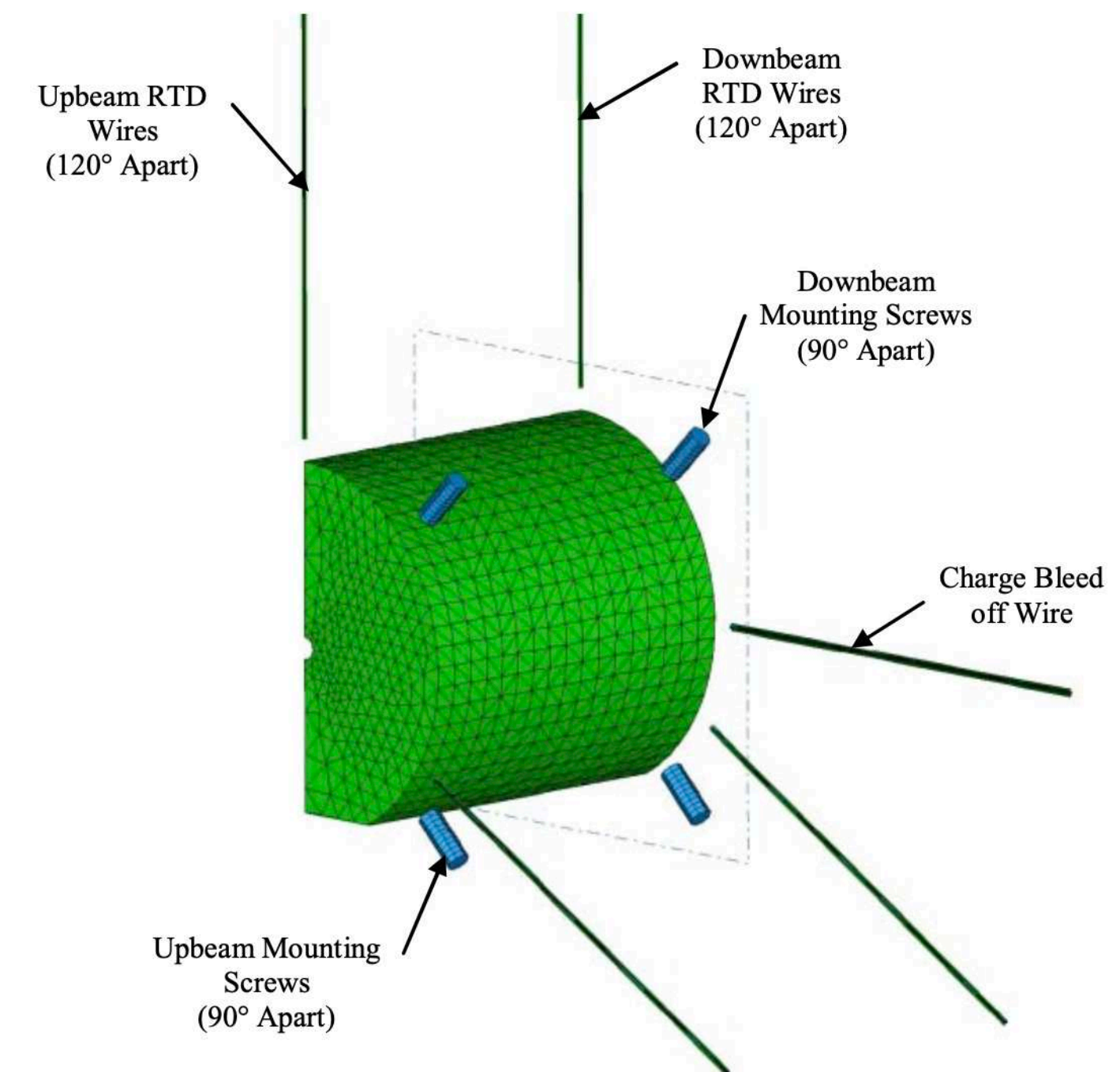
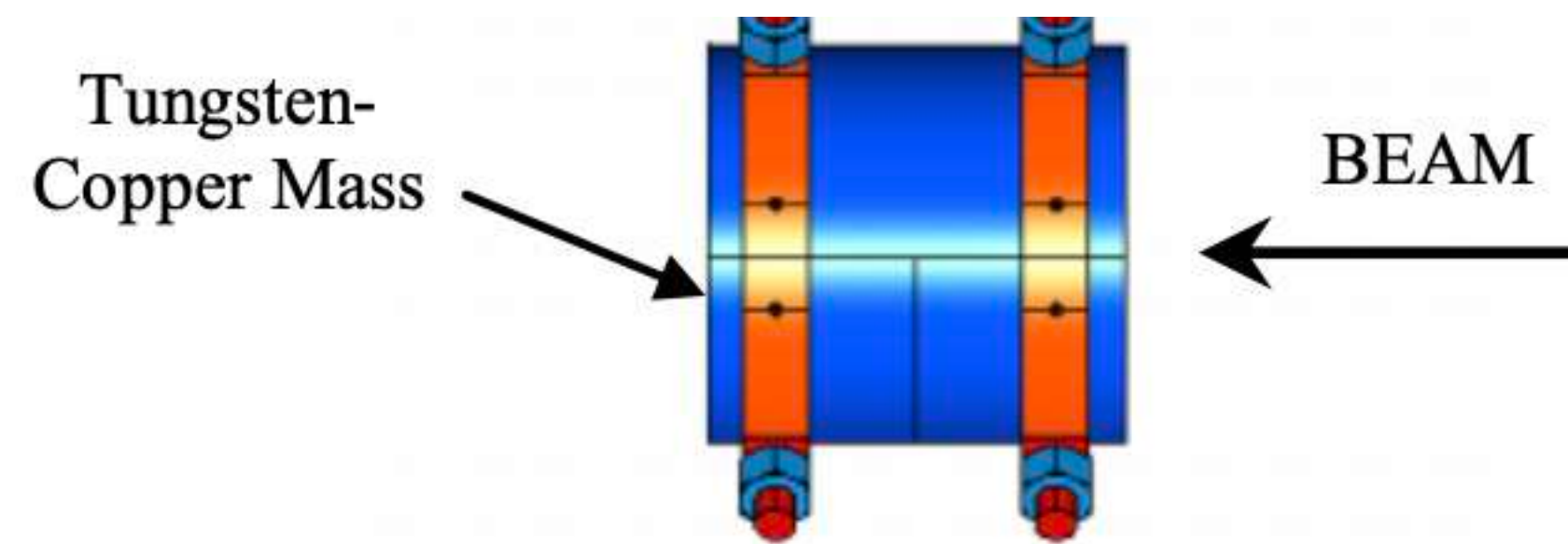


Tungsten-Copper Calorimeter simulation with Geant4

Hector Chinchay

MECHANICAL AND THERMAL DESIGN OF THE CEBAF HALL A BEAM CALORIMETER

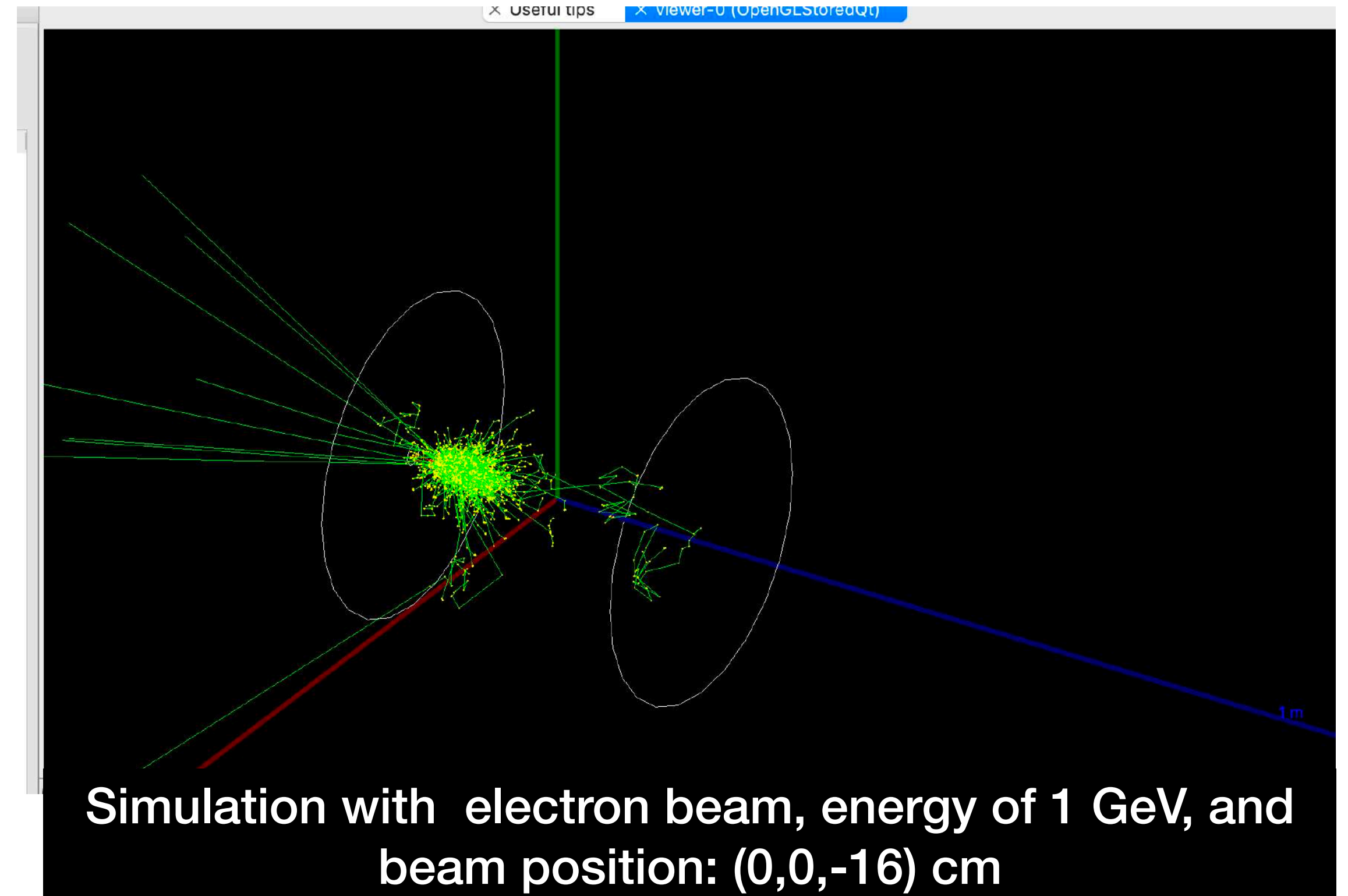
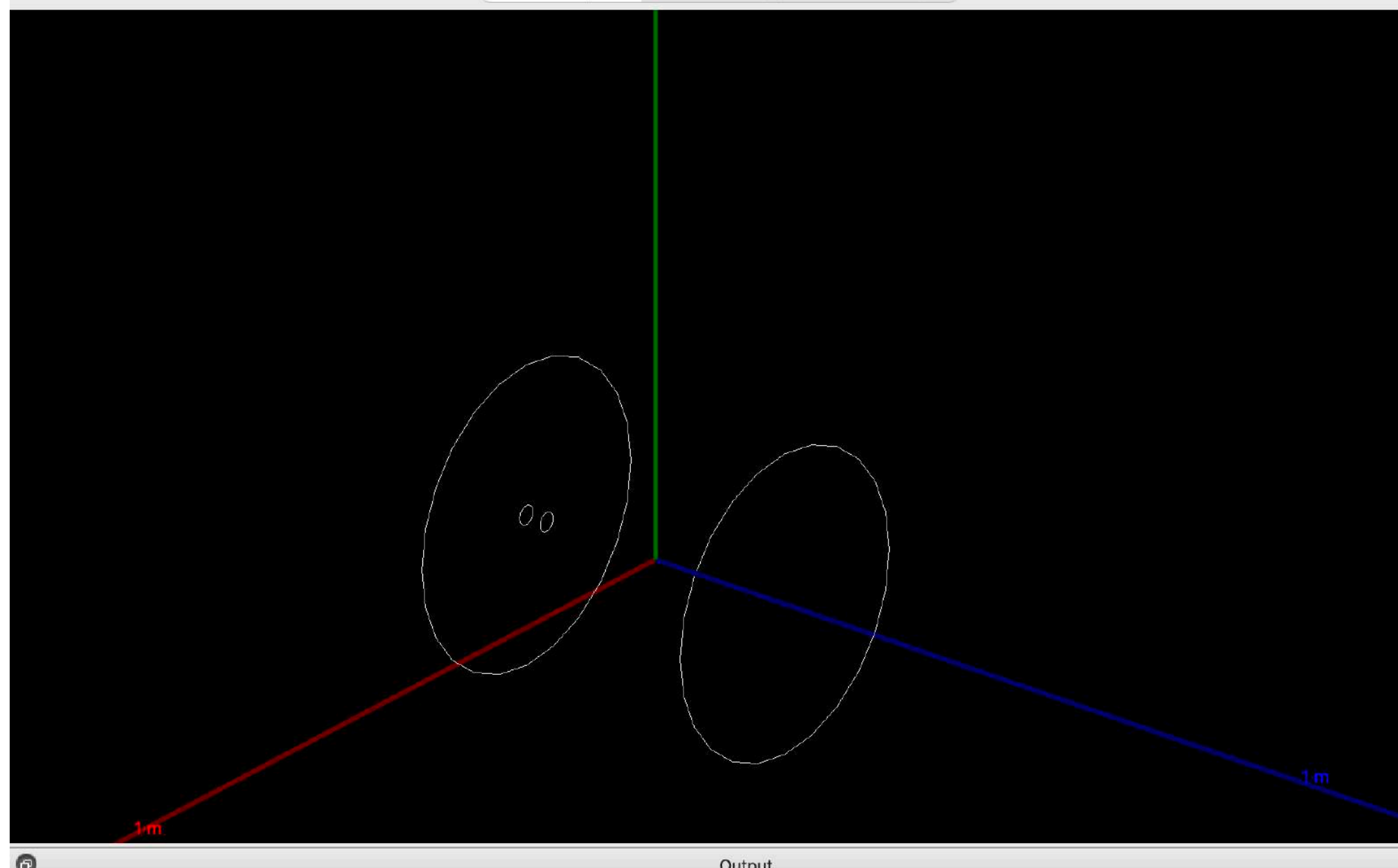
Tungsten- copper (95:5) Calorimeter



16cm diameter by 16cm long tungsten mass with a 1cm diameter by 2.5cm long entrance hole

(Bevins et al., 2006)

Build geometry of calorimeter



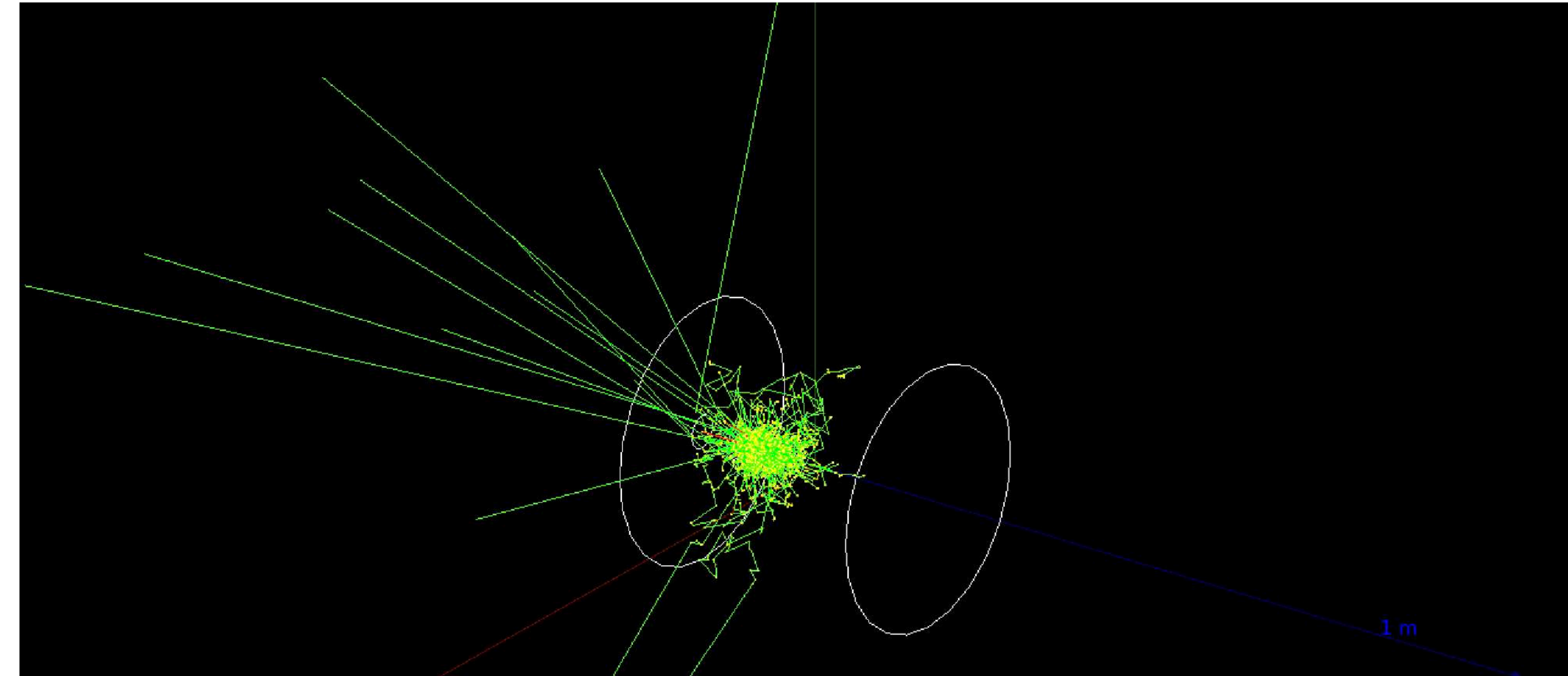
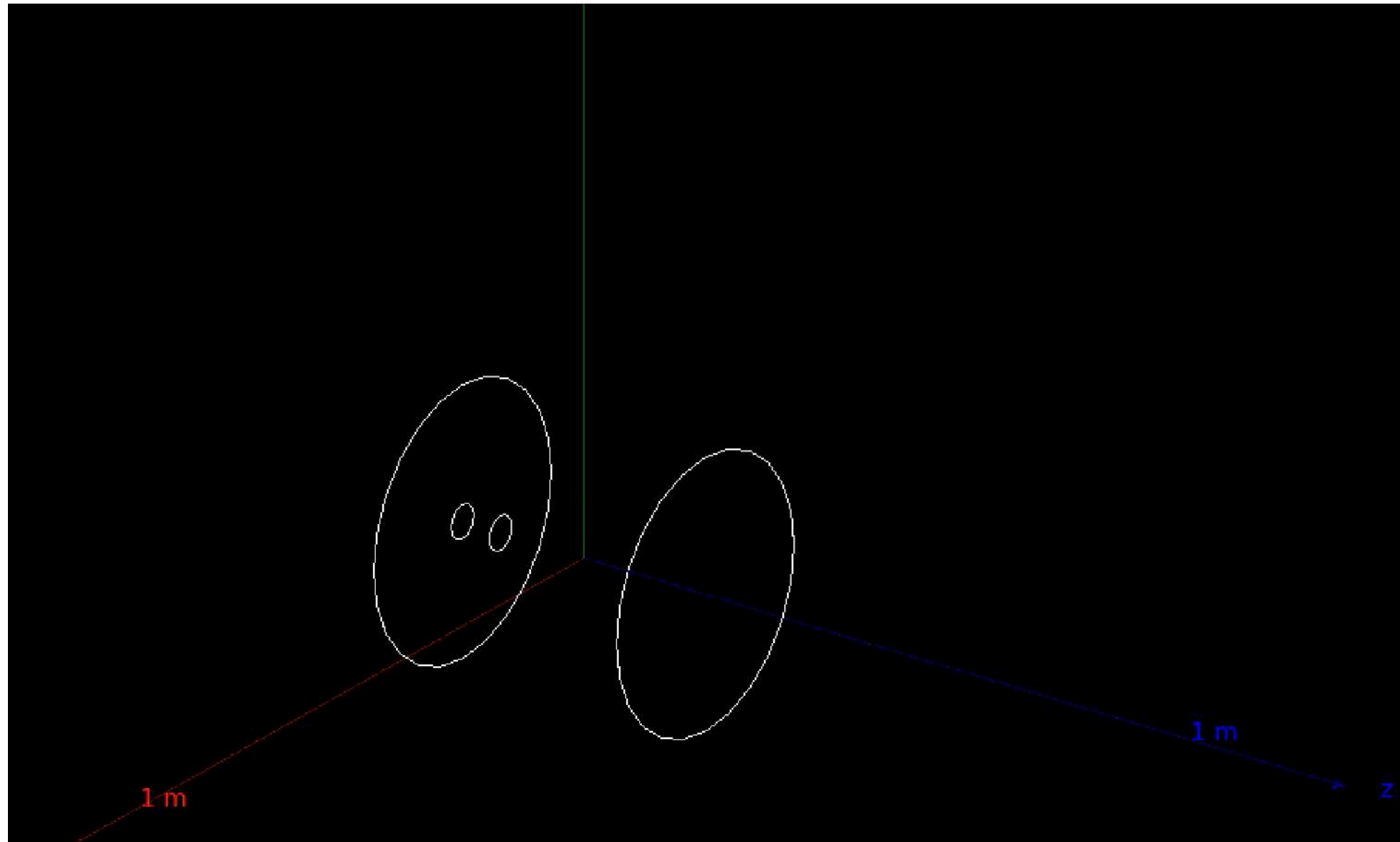
Hole: diameter 1cm, long entrance 2.5cm

```
# User defined icon :  
/gui/addIcon "Run beam on" user_icon "/run/beamOn 1" run.png  
Changing export format to "jpg"  
/gun/particle e-  
/gun/energy 1 GeV  
Matching commands :  
/gun/position  
/gun/polarization  
/gun/position 0 0 -16 cm
```

Session : /run/beamOn 10|

```
G4Material* material = G4Material::GetMaterial("CuW");  
G4Material* material = G4Material::GetMaterial("CsI");  
G4VSolid* Cylinder1  
= new G4Tubs("Cylinder1", 0., 16.*cm, 16.*cm, 0., 2*M_PI*rad);  
G4VSolid* Cylinder2  
= new G4Tubs("Cylinder2", 0., 1.*cm, 2.5*cm, 0., 2*M_PI*rad);  
G4VSolid* hadCalorimeterSolid  
= new G4SubtractionSolid("HadCalorimeterTubs", Cylinder1, Cylinder2, 0, G4ThreeVector(0., 0., -16.*cm));
```

Simulation with electron beam, energy of 1 GeV, and beam position: (0,0,-16) cm
 Hole: diameter 2cm, long entrance 5cm



```

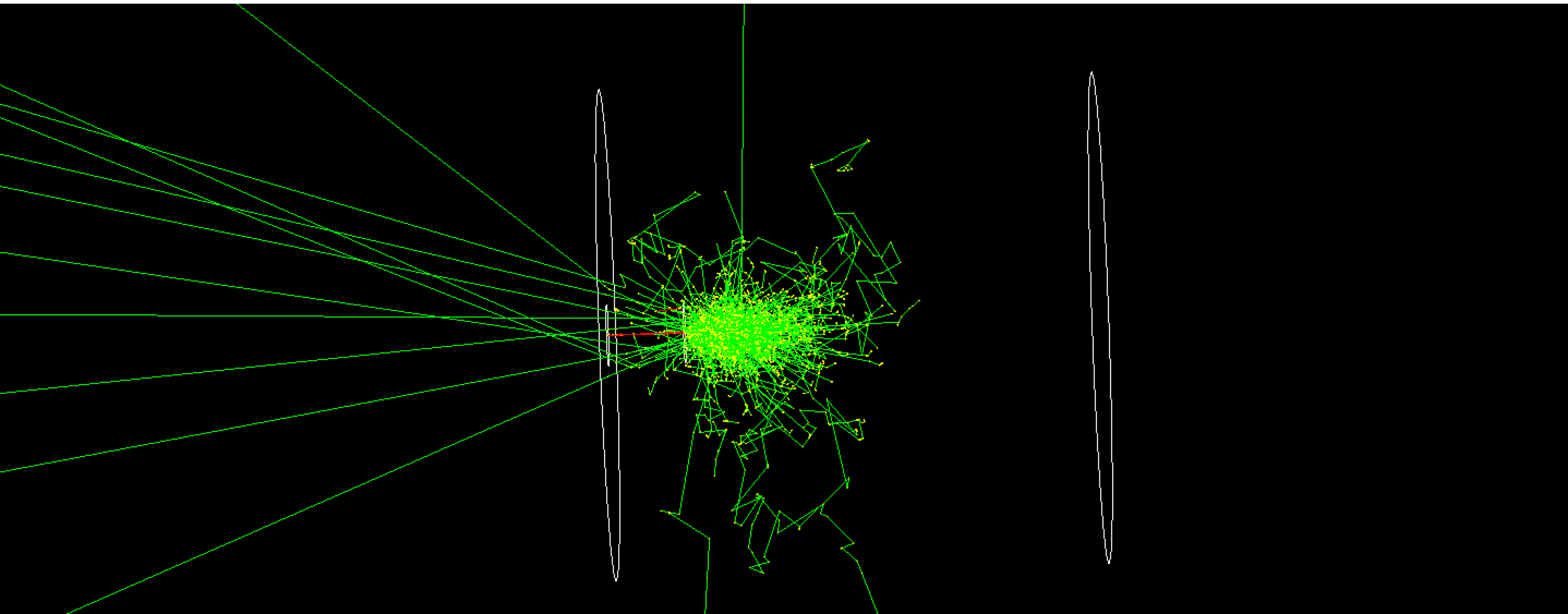
/gui/addButton viewer "Set style wireframe" "/vis/viewer/set/style wireframe"
/gui/addButton viewer "Refresh viewer" "/vis/viewer/refresh"
/gui/addButton viewer "Update viewer (interaction or end-of-file)" "/vis/viewer/update"
/gui/addButton viewer "Flush viewer (= refresh + update)" "/vis/viewer/flush"
/gui/addButton viewer "Update scene" "/vis/scene/notifyHandlers"
#
# User defined icon :
/gui/addIcon "Run beam on" user_icon "/run/beamOn 1" run.png
Changing export format to "jpg"
/gun/particle e-
/gun/position 0 0 -16 cm
/gun/energy 1 GeV
    
```

Session : /run/beamOn 10

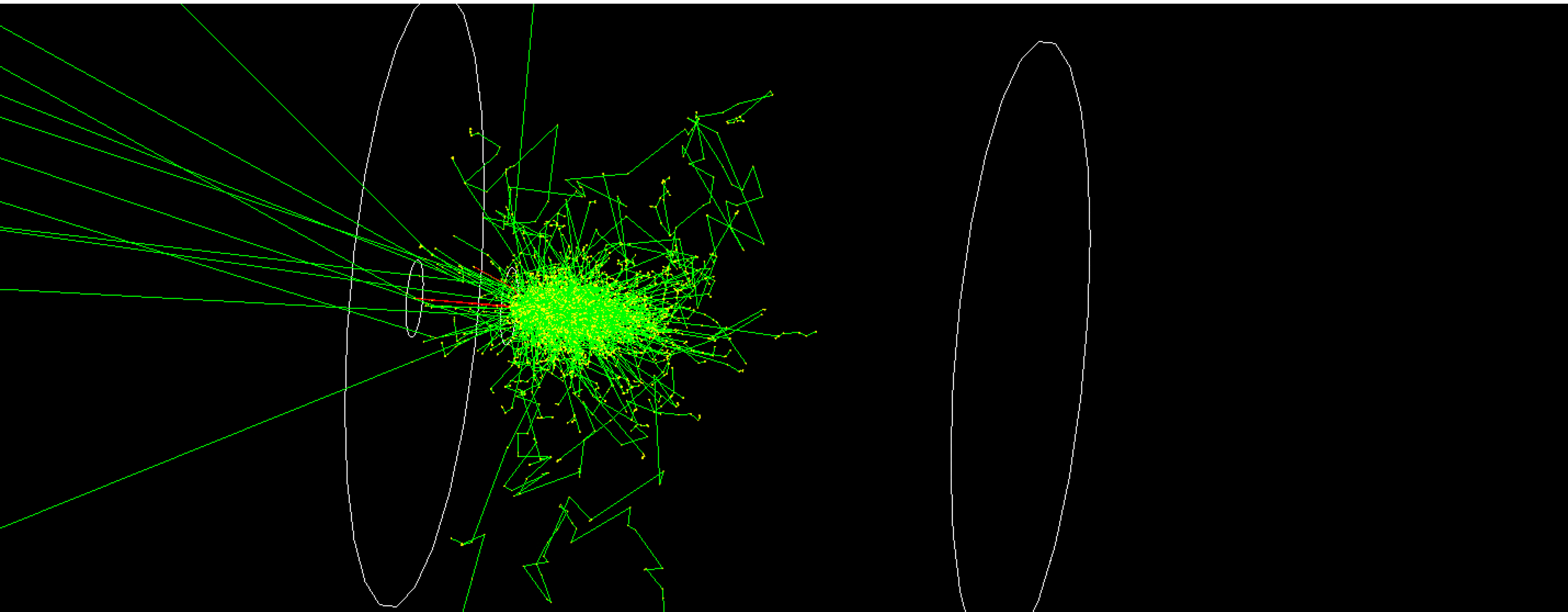
```

// =====
G4Material* material = G4Material::GetMaterial("CuW");
// G4Material* material = G4Material::GetMaterial("CsI");
G4VSolid* Cylinder1
= new G4Tubs("Cylinder1",0.,16.*cm,16.*cm,0.,2*M_PI*rad);
G4VSolid* Cylinder2
= new G4Tubs("Cylinder2",0.,2.*cm,5.*cm,0.,2*M_PI*rad);
G4VSolid* hadCalorimeterSolid
= new G4SubtractionSolid("HadCalorimeterTubs", Cylinder1, Cylinder2, 0, G4ThreeVector(0., 0., -16.*cm));
    
```

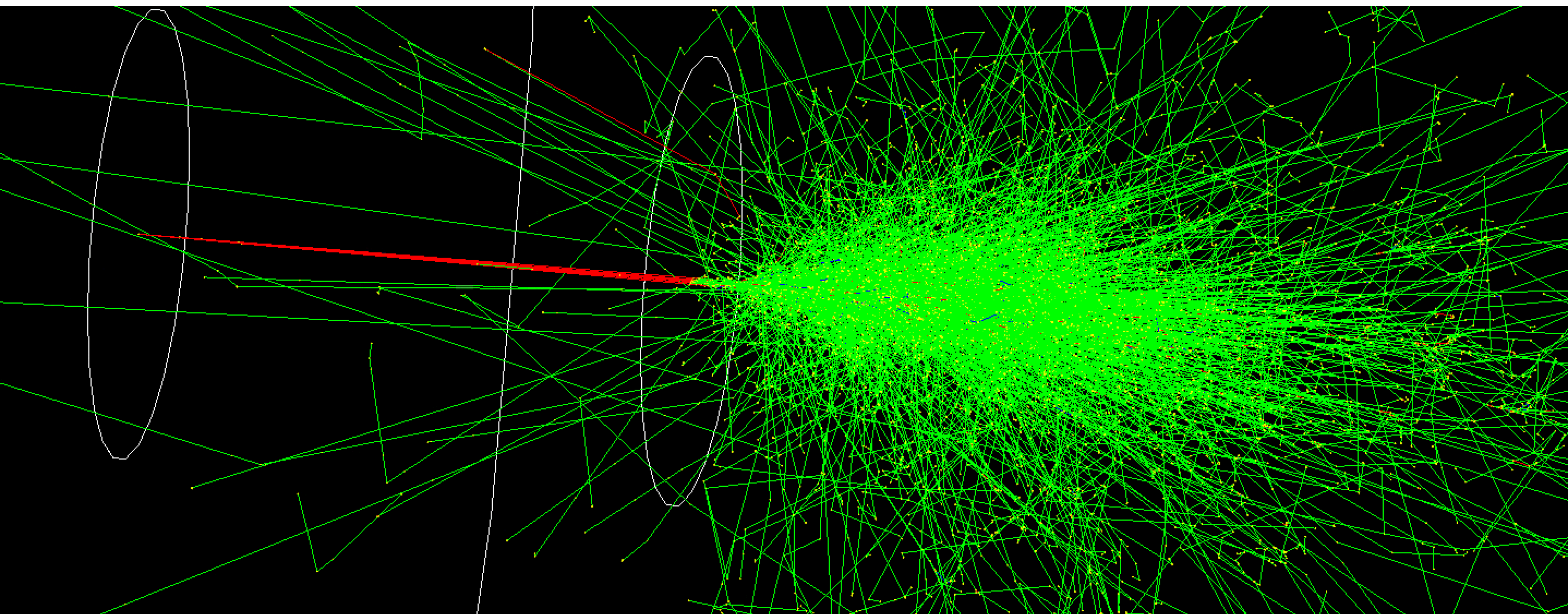
Simulation with electron beam, energy of 1 GeV, and beam position: (0,0,-16) cm
Hole: diameter 2cm, long entrance 5cm



Simulation with electron beam, energy of 1 GeV, and beam position: (0,0,-16) cm
Hole: diameter 2cm, long entrance 5cm



Zoom in the hole



Define material of calorimeter

Materials:
CuW
Air
Air_lowDensity

Material: CuW density: 16.750 g/cm3 RadL: 4.238 mm Nucl.Int.Length: 11.396 cm
Imean: 663.705 eV temperature: 293.15 K pressure: 1.00 atm

```

----> Element: Copper (Cu)  Z = 29.0  N = 64  A = 63.550 g/mole
----> Isotope: Cu63  Z = 29  N = 63  A = 62.93 g/mole  abundance: 69.170 %
----> Isotope: Cu65  Z = 29  N = 65  A = 64.93 g/mole  abundance: 30.830 %
      ElmMassFraction: 10.00 %  ElmAbundance 24.32 %

----> Element: Tungsten (W)  Z = 74.0  N = 184  A = 183.800 g/mole
----> Isotope: W180  Z = 74  N = 180  A = 179.95 g/mole  abundance: 0.120 %
----> Isotope: W182  Z = 74  N = 182  A = 181.95 g/mole  abundance: 26.500 %
----> Isotope: W183  Z = 74  N = 183  A = 182.95 g/mole  abundance: 14.310 %
----> Isotope: W184  Z = 74  N = 184  A = 183.95 g/mole  abundance: 30.640 %
----> Isotope: W186  Z = 74  N = 186  A = 185.95 g/mole  abundance: 28.430 %
      ElmMassFraction: 90.00 %  ElmAbundance 75.68 %
    
```

/material/g4/printMaterial

Material: G4_AIR density: 1.205 mg/cm3 RadL: 303.921 m Nucl.Int.Length: 710.095 m
Imean: 85.700 eV temperature: 293.15 K pressure: 1.00 atm

```

----> Element: C (C)  Z = 6.0  N = 12  A = 12.011 g/mole
----> Isotope: C12  Z = 6  N = 12  A = 12.00 g/mole  abundance: 98.930 %
----> Isotope: C13  Z = 6  N = 13  A = 13.00 g/mole  abundance: 1.070 %
      ElmMassFraction: 0.01 %  ElmAbundance 0.02 %

----> Element: N (N)  Z = 7.0  N = 14  A = 14.007 g/mole
----> Isotope: N14  Z = 7  N = 14  A = 14.00 g/mole  abundance: 99.632 %
----> Isotope: N15  Z = 7  N = 15  A = 15.00 g/mole  abundance: 0.368 %
      ElmMassFraction: 75.53 %  ElmAbundance 78.44 %

----> Element: O (O)  Z = 8.0  N = 16  A = 15.999 g/mole
----> Isotope: O16  Z = 8  N = 16  A = 15.99 g/mole  abundance: 99.757 %
----> Isotope: O17  Z = 8  N = 17  A = 17.00 g/mole  abundance: 0.038 %
----> Isotope: O18  Z = 8  N = 18  A = 18.00 g/mole  abundance: 0.205 %
      ElmMassFraction: 23.18 %  ElmAbundance 21.07 %

----> Element: Ar (Ar)  Z = 18.0  N = 40  A = 39.948 g/mole
----> Isotope: Ar36  Z = 18  N = 36  A = 35.97 g/mole  abundance: 0.337 %
----> Isotope: Ar38  Z = 18  N = 38  A = 37.96 g/mole  abundance: 0.063 %
----> Isotope: Ar40  Z = 18  N = 40  A = 39.96 g/mole  abundance: 99.600 %
      ElmMassFraction: 1.28 %  ElmAbundance 0.47 %
    
```

Material: Air_lowDensity density: 0.000 mg/cm3 RadL: 30392.070 km Nucl.Int.Length: 71009.501
Imean: 85.700 eV temperature: 293.15 K pressure: 1.00 atm

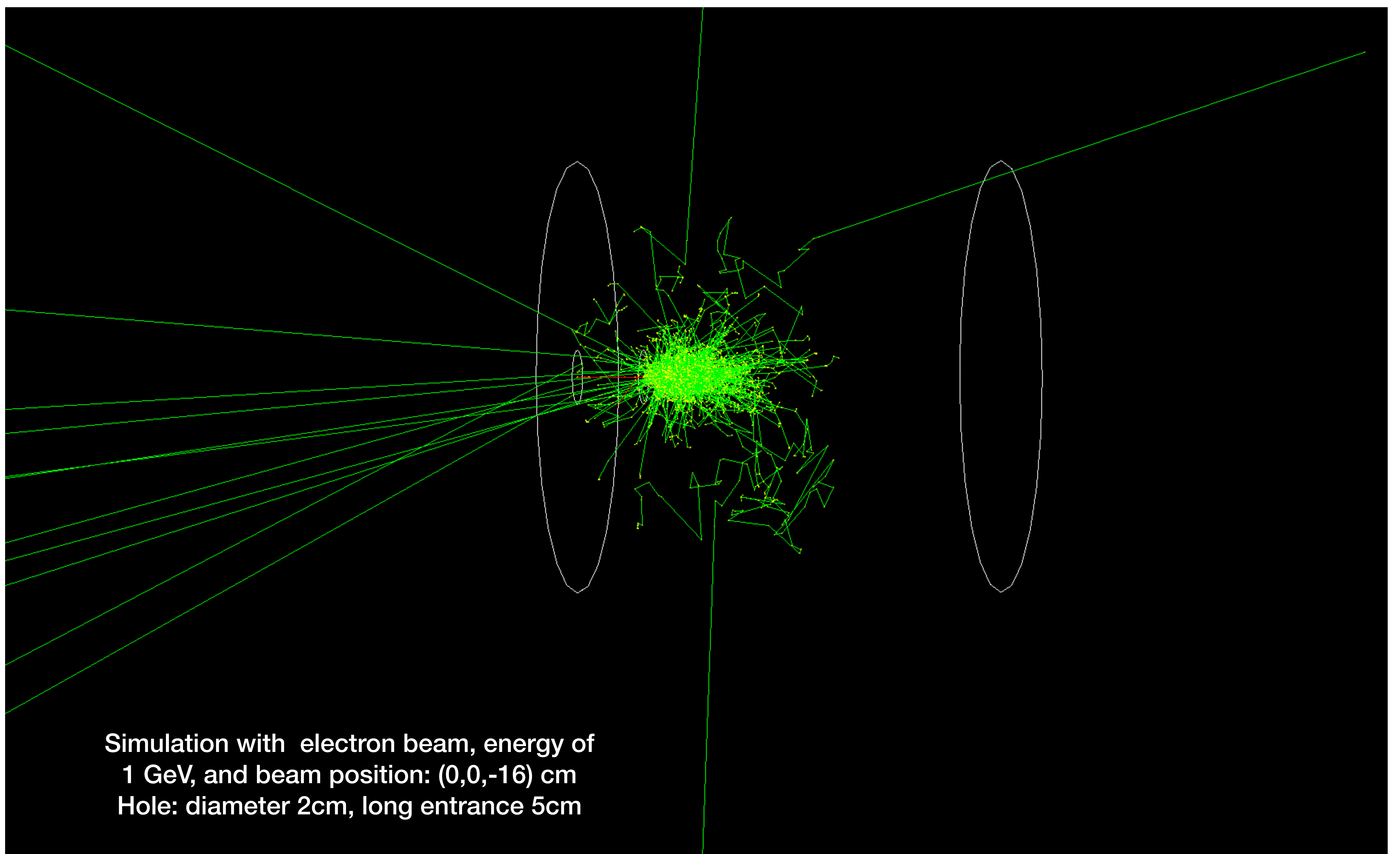
```

----> Element: C (C)  Z = 6.0  N = 12  A = 12.011 g/mole
----> Isotope: C12  Z = 6  N = 12  A = 12.00 g/mole  abundance: 98.930 %
----> Isotope: C13  Z = 6  N = 13  A = 13.00 g/mole  abundance: 1.070 %
      ElmMassFraction: 0.01 %  ElmAbundance 0.02 %

----> Element: N (N)  Z = 7.0  N = 14  A = 14.007 g/mole
----> Isotope: N14  Z = 7  N = 14  A = 14.00 g/mole  abundance: 99.632 %
----> Isotope: N15  Z = 7  N = 15  A = 15.00 g/mole  abundance: 0.368 %
      ElmMassFraction: 75.53 %  ElmAbundance 78.44 %

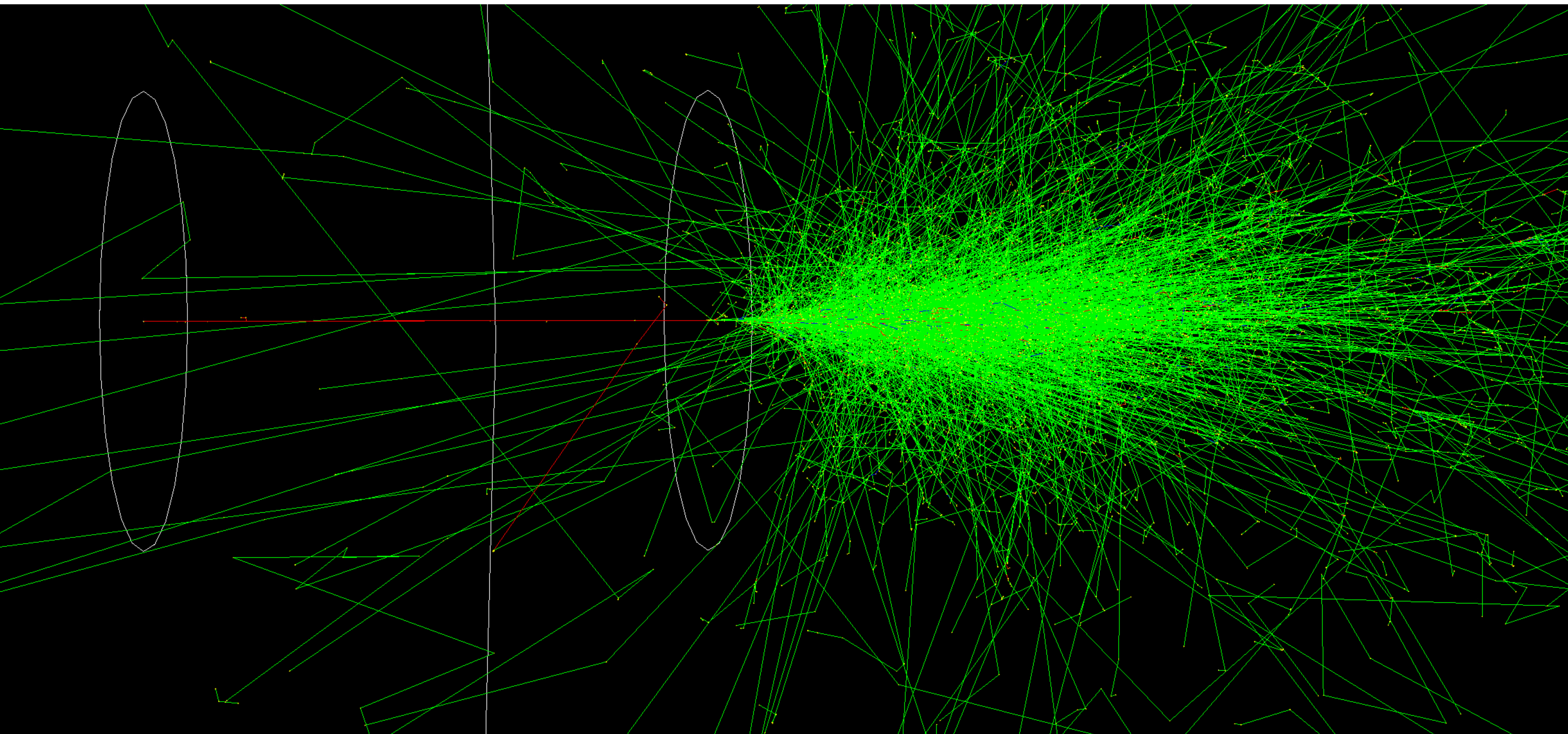
----> Element: O (O)  Z = 8.0  N = 16  A = 15.999 g/mole
----> Isotope: O16  Z = 8  N = 16  A = 15.99 g/mole  abundance: 99.757 %
----> Isotope: O17  Z = 8  N = 17  A = 17.00 g/mole  abundance: 0.038 %
----> Isotope: O18  Z = 8  N = 18  A = 18.00 g/mole  abundance: 0.205 %
      ElmMassFraction: 23.18 %  ElmAbundance 21.07 %

----> Element: Ar (Ar)  Z = 18.0  N = 40  A = 39.948 g/mole
----> Isotope: Ar36  Z = 18  N = 36  A = 35.97 g/mole  abundance: 0.337 %
----> Isotope: Ar38  Z = 18  N = 38  A = 37.96 g/mole  abundance: 0.063 %
----> Isotope: Ar40  Z = 18  N = 40  A = 39.96 g/mole  abundance: 99.600 %
      ElmMassFraction: 1.28 %  ElmAbundance 0.47 %
    
```

Simulation with electron beam, energy of
1 GeV, and beam position: (0,0,-16) cm
Hole: diameter 2cm, long entrance 5cm

Zoom in the hole



Size of electron beam on target

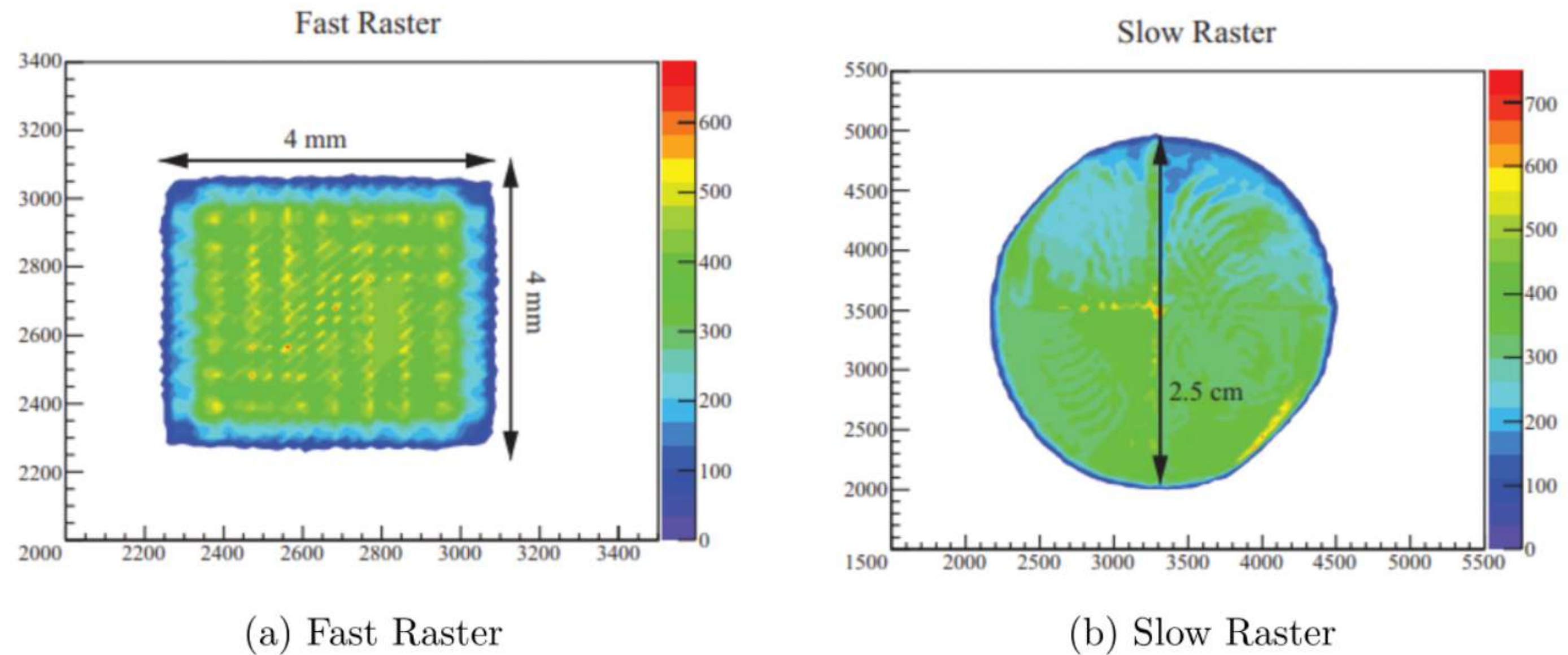
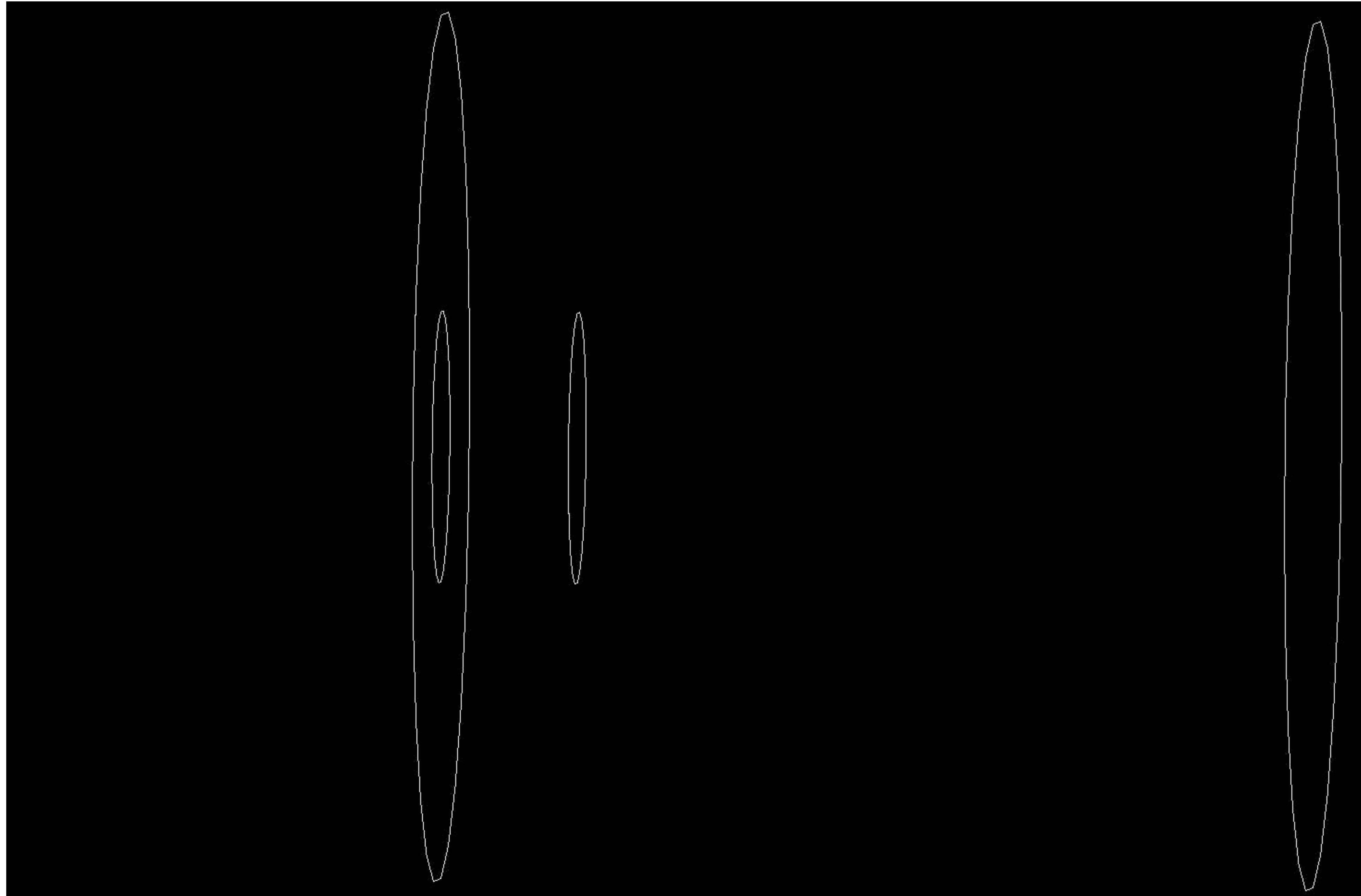
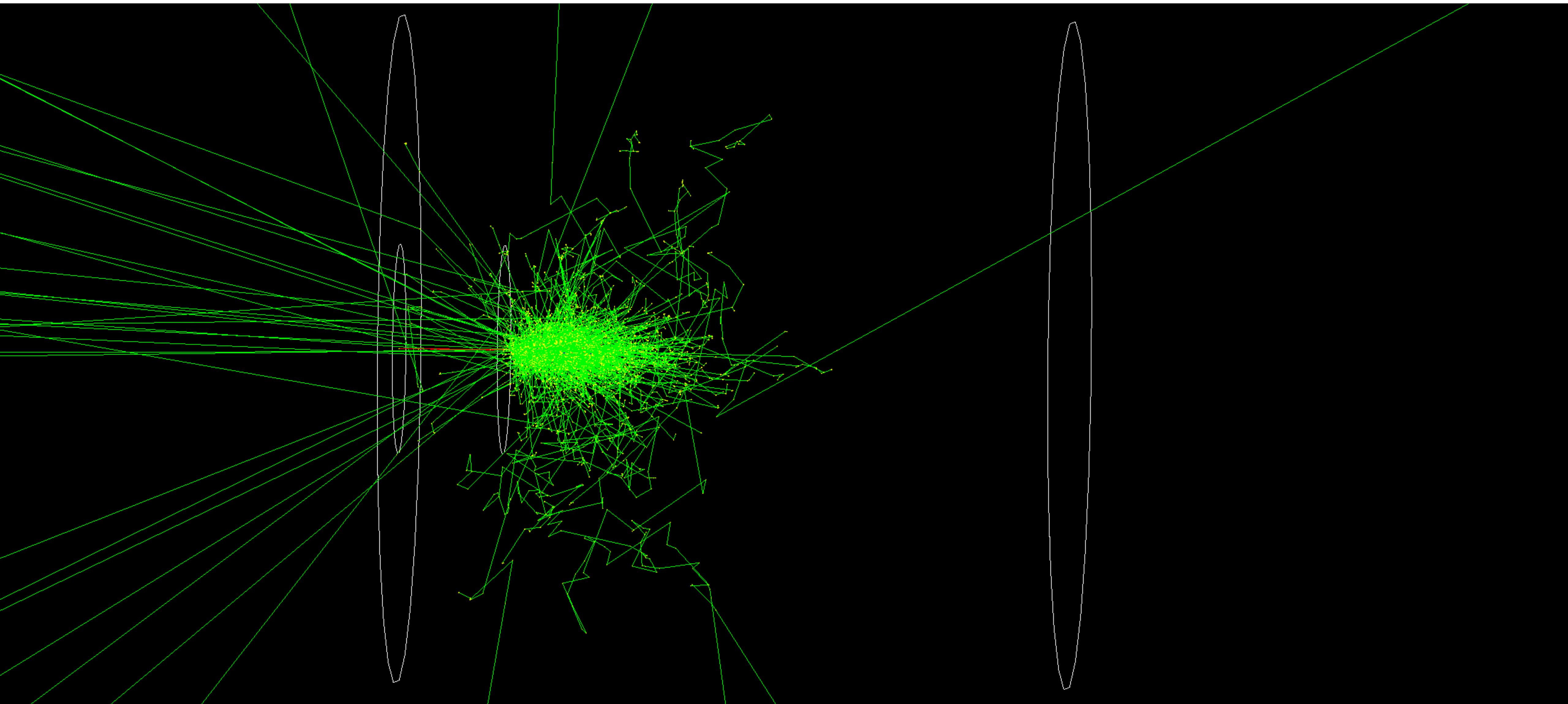


Figure 4.4: Raster swept beam profiles. Plots are a function of current in arbitrary units. Reproduced from Zielinski [2017]

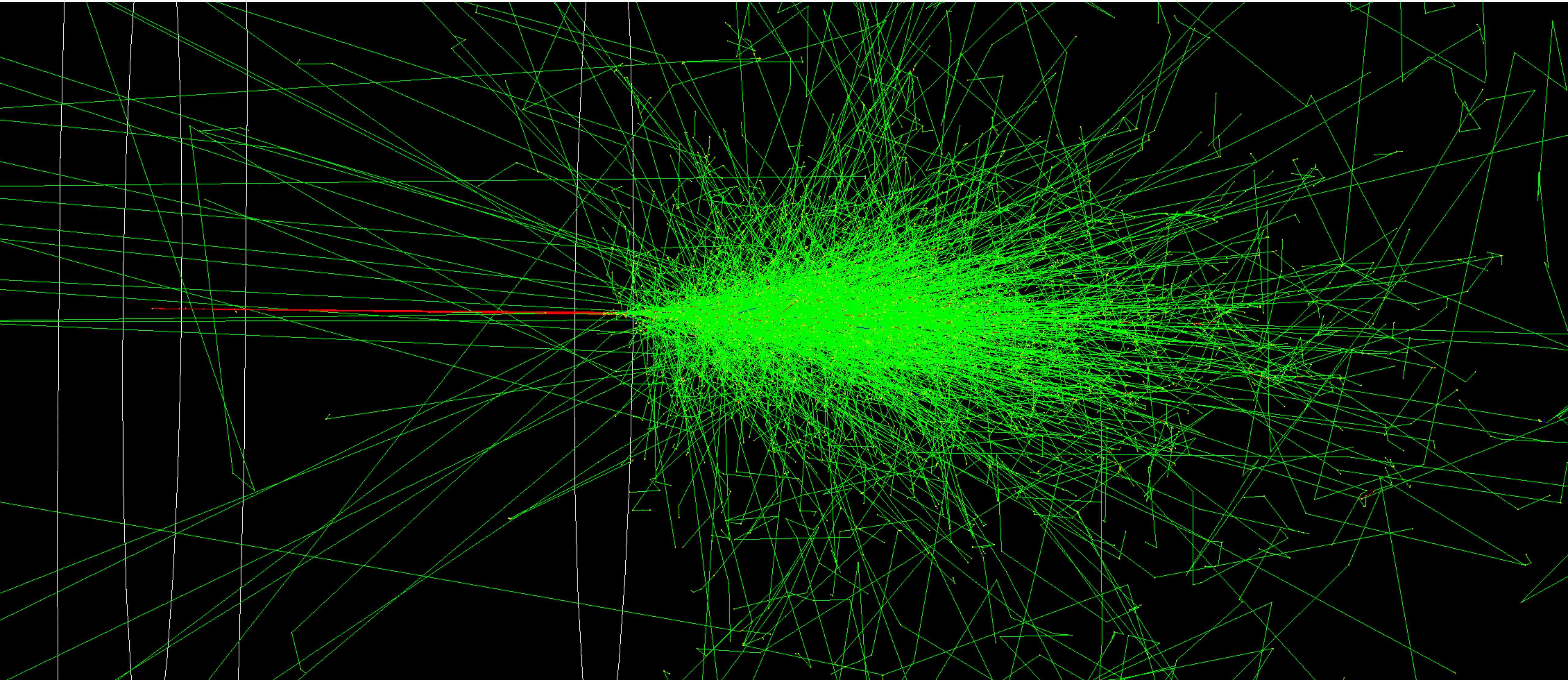
Hole: diameter 5cm, long entrance 5cm



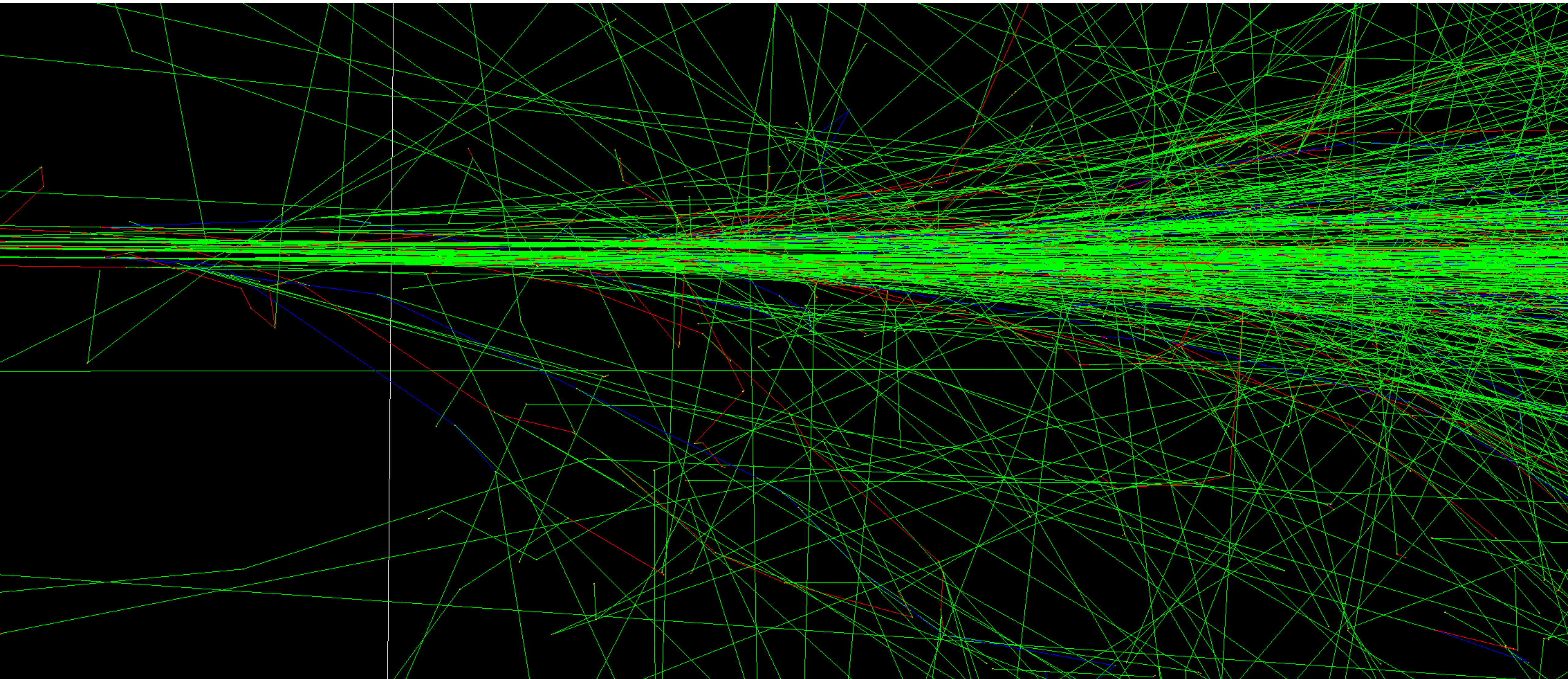
Simulation with electron beam, energy of 1 GeV, and beam position: (0,0,-16) cm
Hole: diameter 5cm, long entrance 5cm



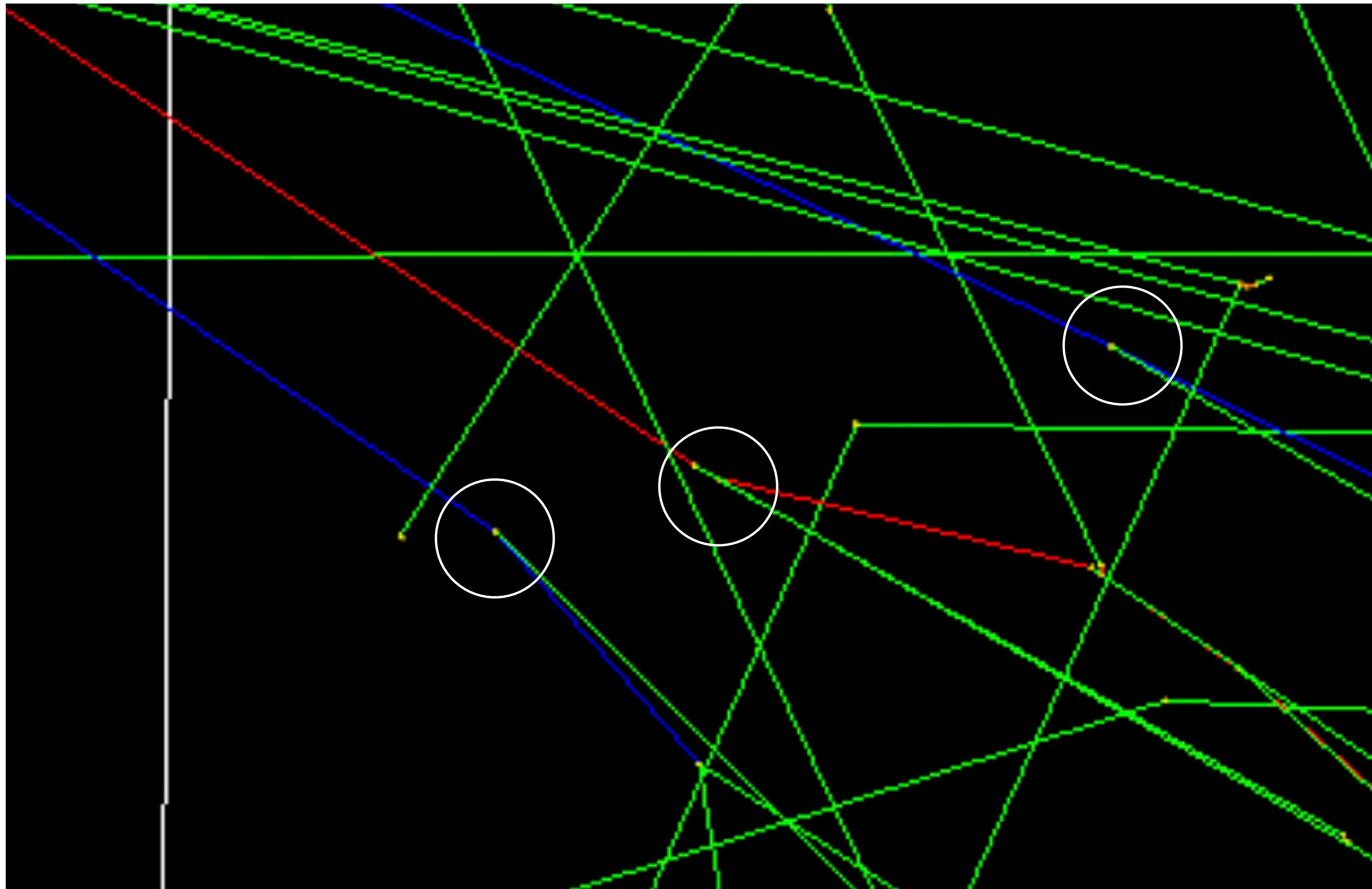
Zoom in the hole



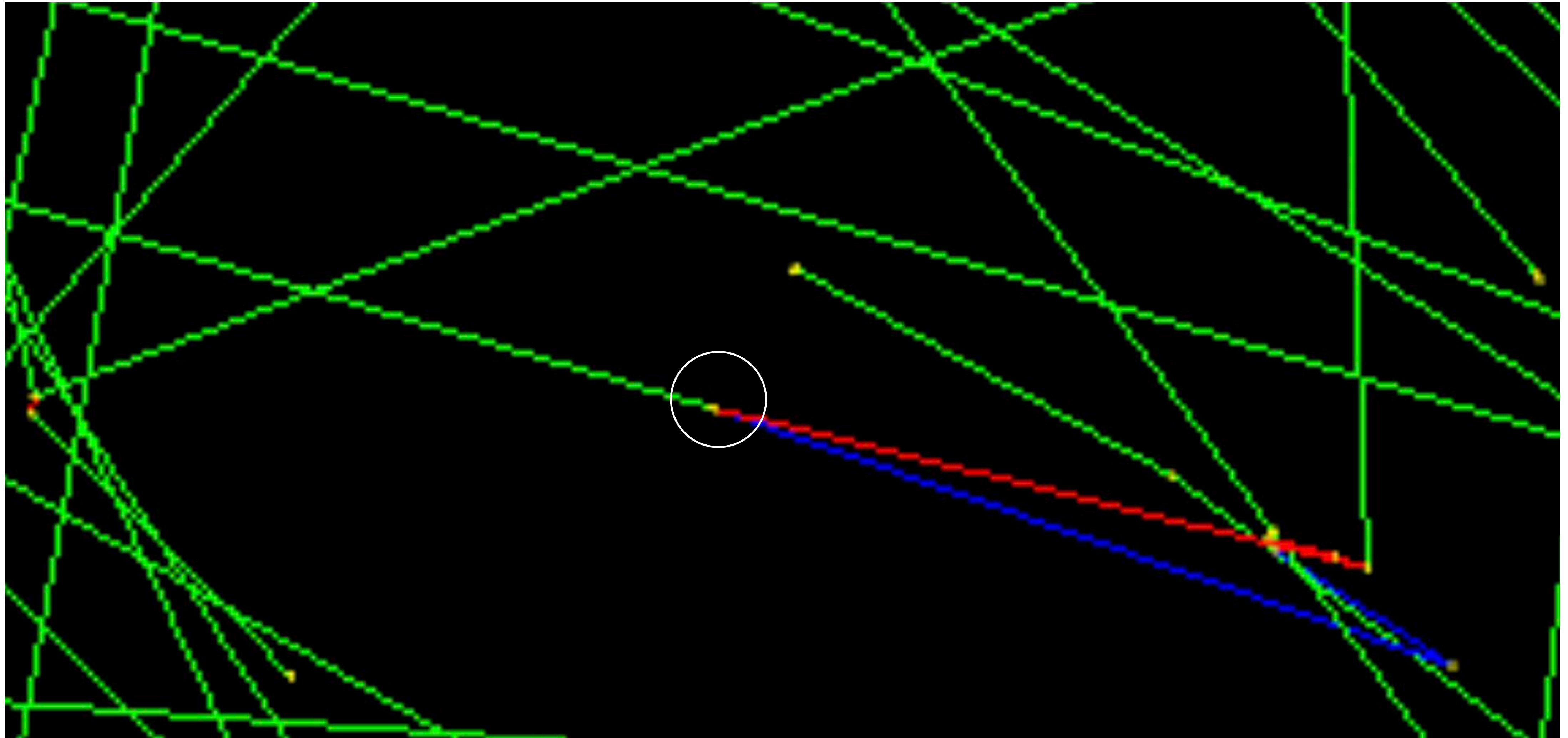
Zoom in the hole



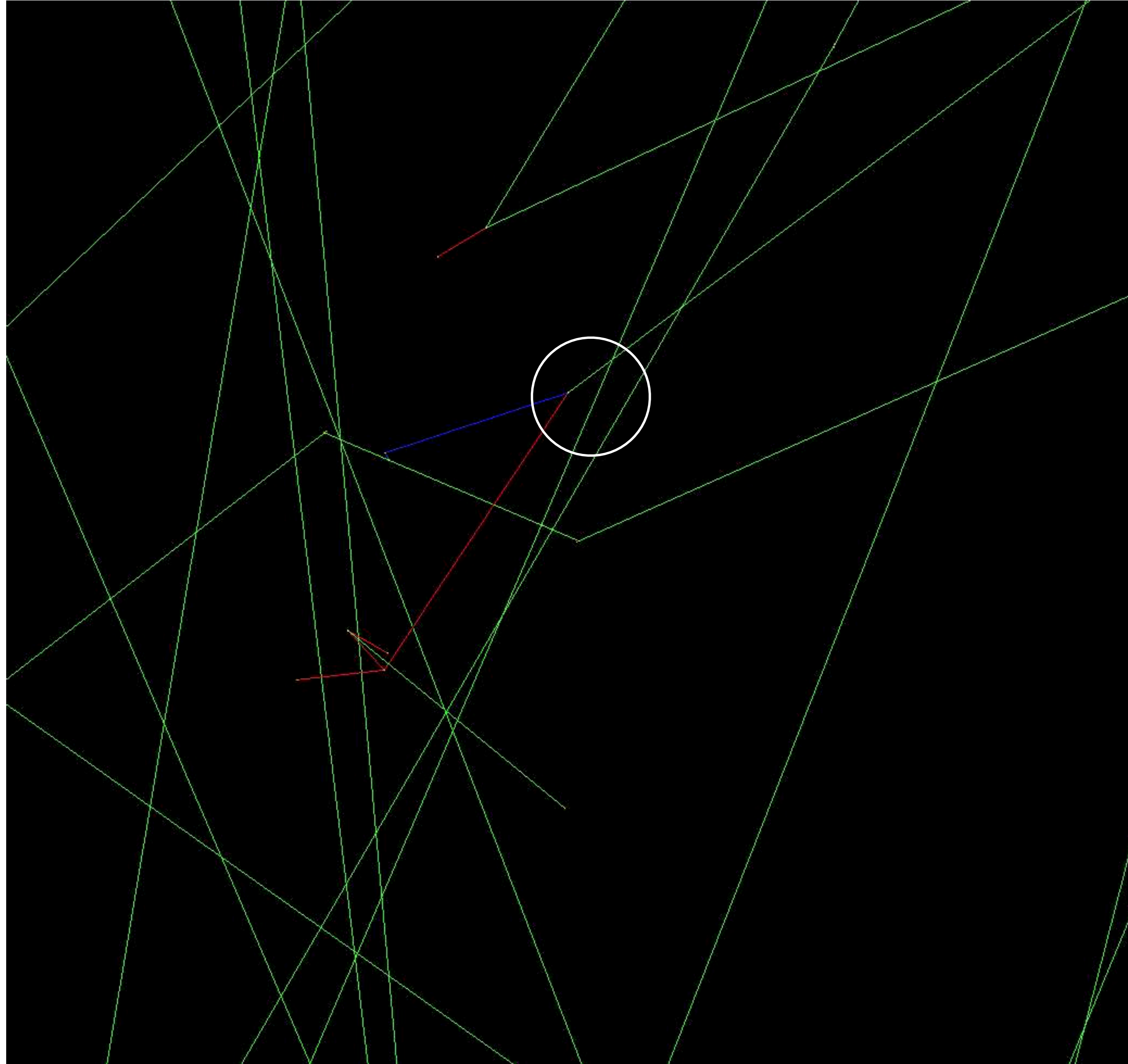
Electrons (**Positrons**) can accelerate (change direction of motion) by absorbing or emitting a **photon**.



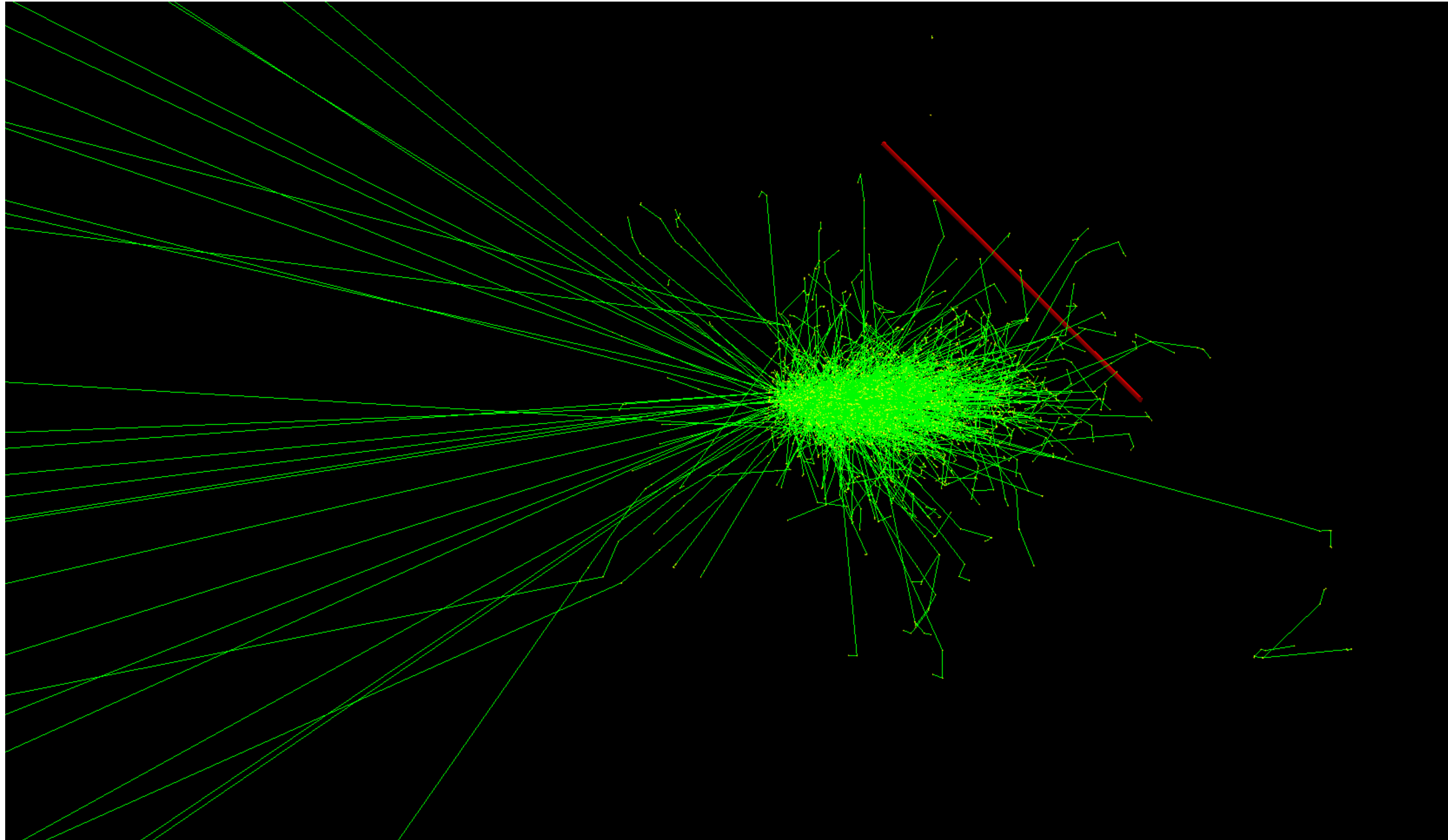
Electron-Positron Pair production from a photon (γ)



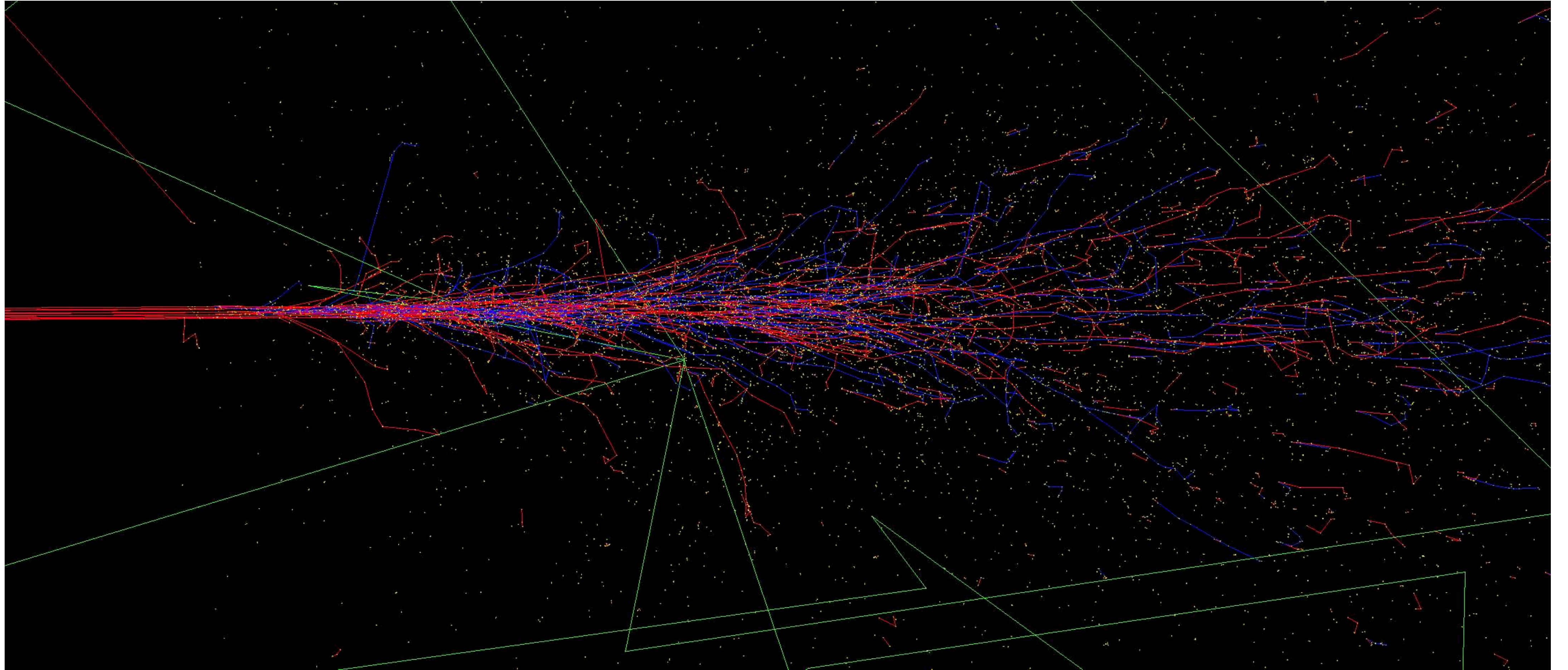
Annihilation of an **electron** and **positron** into a **photon** (γ)



Filtering trajectories: Only allows gammas



Filtering trajectories: Only allows electrons/positrons



Next step: Reproduction of dE/dx vs Z (cm) and dE/dx vs X (cm) and plots

Related G4 Simulations by Nathaly Santiesteban

